



## **ENVIRONMENT AUDIT REPORT**

**CONSULTATION REPORT** 



## KURUKSHETRA UNIVERSITY, KURUKSHETRA

(Established by the State legislature act XII of 1956) (A+ Grade NAAC Accredited)

Thanesar, Haryana 136119

PREPARED BY

#### EMPIRICAL EXERGY PRIVATE LIMITED

Flat No. 201, OM Apartment,214 IndrapuriColony, Bhawarkuan,Indore – 452 001 (M. P.), India 0731-4948831, 7869327256 Email ID:eempirical18@gmail.com www.eeplgroups.com (2020-21)





### **CONTENT**

Items	Page No
ACKNOWLEDGEMENT	3
ACCREDITATION CERTIFICATE	4
EXECUTIVE SUMMARY	5
Introduction	8
About University	8
About University Campus	10
Environmental Monitoring Committee	15
The Audit Team	15
About Water Auditing	16
Objective of Environment Audit	16
Target area of Environment audit	16
Methodology Followed for conducting Environment Audit	17
Water Consumption and waste water sources	18
Details of source fresh water and uses area	18
Water Accounting and metering system	19
Water Storage Capacity in University campus	20
Water uses area in University Campus	24
Fresh Water uses for Gardening:	34
Waste water generation sources of University campus	36
Water Treatment Plant in KUK	37
On Site water flow measurement in KUK.	43
Water test report of ground water in KUK University	45
Waste Management	49
About Waste	49
Waste management Practices adopted by the University	50
Waste Collection Points in University	51
Organic Waste Generation in University	52
Rain Water Harvesting System	55
About rain water harvesting	55
Estimated Rain water harvesting Potential of the University	56
Design of RWH in KUK	57
Photograph of RWH in KUK	58
	ACKNOWLEDGEMENT ACCREDITATION CERTIFICATE EXECUTIVE SUMMARY Introduction About University About University Campus Environmental Monitoring Committee The Audit Team About Water Auditing Objective of Environment Audit Target area of Environment audit Methodology Followed for conducting Environment Audit Water Consumption and waste water sources Details of source fresh water and uses area Water Accounting and metering system Water Storage Capacity in University campus Water uses area in University Campus Fresh Water uses for Gardening: Waste water generation sources of University campus Water Treatment Plant in KUK On Site water flow measurement in KUK . Water test report of ground water in KUK University Waste Management About Waste Waste management Practices adopted by the University Waste Collection Points in University Organic Waste Generation in University Rain Water Harvesting System About rain water harvesting Estimated Rain water harvesting Potential of the University Design of RWH in KUK





## **ACKNOWLEDGEMENT**

**Empirical Exergy Private Limited (EEPL), Indore (M.P)**takes this opportunity to appreciate & thank the management of **Kurukshetra University Kurukshetra Haryana.** for giving us an opportunity to conduct environment audit for the University.

We are indeed touched by the helpful attitude and co-operation of all faculties and technical staff, who rendered their valuable assistance and co-operation the course of study.



Rajesh Kumar Singadiya

(Director)

M.Tech (Energy Management), PhD (Research Scholar)

Accredited Energy Auditor [AEA-0284]

Certified Energy Auditor [CEA-7271]

(BEE, Ministry of Power, Govt. of India)

Empanelled Energy Auditor with MPUVN, Bhopal M.P.

Lead Auditor ISO50001:2011 [EnMS) from FICCI, Delhi

Certified Water Auditor (NPC, Govt of India)

Charted Engineer [M-1699118], The Institution of Engineers (India)

Member of ISHRAE [58150]







## **BUREAU OF ENERGY EFFICIENCY**

Examination Registration No.: EA-7271

Accreditation Registration No.: AEA-284



## Certificate of Accreditation

The certificate is subject to the provisions of the Bureau of Energy Efficiency (Qualifications for Accredited Energy Auditors and Maintenance of their List) Regulations, 2010.

This certificate shall be valid until it is cancelled under regulation 9 of the Bureau of Energy Efficiency (Qualifications for Accredited Energy Auditors and Maintenance of their List) Regulations, 2010.

On cancellation, the certificate of accreditation shall be surrendered to the Bureau within fifteen days from the date of receipt of order of cancellation.

Your name has been entered at AEA No....284.... in the register of list of accredited energy auditors. Your name shall be liable to be struck out on the grounds specified in regulation 8 of the Bureau of Energy Efficiency (Qualifications for Accredited Energy Auditors and Maintenance of their List) Regulations, 2010.

Given under the seal of the Bureau of Energy Efficiency, Ministry of Power, this 5th day of October, 2018

Secretary, Bureau of Energy Efficiency New Delhi







## **EXECUTIVE SUMMARY**

The executive summary of the Environmentaudit report furnished in this section briefly gives the identified water conservation measures, that can be implemented in a phased manner to conserve the water and increase the productivity of the University.

### **Initiative for EnvironmentManagement Taken by University**

#### **RAINWATER HARVESTING SYSTEM:**

- ❖ University has 14 no's of "Rainwater harvesting system" in University campus for maintaining ground water label. This system saves about 80 to 90 % of roof top rain water of the building. It is Appreciable.
- ❖ Audit team discussed with KUK Management. They informed that University has proposed 34 nos New Rain water harvesting pit. It is under process. It is Appreciable.

#### **SPRINKLE IRRIGATION SYSTEM FOR LAWN**

❖ University uses sprinkle irrigation system for Lawn area. It is Appreciable.

#### **WASTE WATER MANAGEMENT:**

❖ University has 2.4 MLD STP plant for waste water treatment. The waste water generated from various departments, Hostels, and from residential area. It is Appreciable

#### **L-WASTE MANAGEMENT: -**

❖ University has E-waste management system for the final safe disposal of the E-waste. The list of e-waste items written off by the write off committee and approved by concerned authority is provided to co-ordinator, E-Waste (Head Computer Centre) for further disposal to E-waste vendor according to E-Waste policy.

## AREAS FOR IMPROVEMENT

#### **WATER MONITORING SYSTEM:**

❖ Installation of "Cloud based (IoT based) Ground Water extraction monitoring system" for bore-well to quantify fresh water consumption per day in the University".

#### **♣** WASTE WATER MEASUREMENT

❖ Installation of "Water flow meter" on STP Plant to measure treated waste water per day. It will also be helpful for determination of chemical & operational cost of the plants.





#### **♣** REPLACEMENT OF ALL OLD BOREWELL STARTERS: -

❖ It is recommended to update old electrical starter panels by new updated System.

#### **♣** DRIP WATER IRRIGATION SYSTEM FOR PLANTS.

- Use drip water irrigation system for Plants.
- **USE SENSOR BASED EFFICIENT WATER TAPS: -**
- ❖ Water saving taps either reduce water flow or automatically switch off to help save water. So, it is highly recommended to install efficient water taps in the University campus to reduce water wastage.

#### **↓** USE SENSOR BASED EFFICIENT URINAL TAPS: -

❖ Replacing these inefficient fixtures with water sensor labelled flushing urinal can save between 0.5 to 04 litters per flush without sacrificing performance. Installing water saving flushing urinal will not only reduce water use in facilities but also save money on water bills.

#### **♣ INSTALL ETP PLANT: -**

❖ University has 9 no's of departments that use chemicals in lab activity and generate waste water also. It is highly recommended for all the 9 departments to install separate drain line and install ETP plant for treatment of contaminated water.

#### **SOLID WASTE MANAGEMENT:**

❖ The basic principle of good waste management practice is based on the concept of 3R's namely, reduce, recycle, and reuse.

#### **ADOPT 03 DUST BIN SYSTEM: -**

❖ It is highly recommended to adopt 03 dust bin system in University. At present University has single dust bin system.

#### **↓** INSTALL ORGANIC CONVERTER: -

There is good potential to install organic converter in hostels for kitchen and vegetable organic waste.

#### **VERMICOMPOST UNIT:**

❖ Install vermicompost unit for Biodegradable waste generated from garden and lawn area.





#### OTHER SUGGESTIONS.

Some very important suggestions are: -

- ♣ Prepare the water management policy and work towards creating and implementing a strategy to reduce the water consumption.
- ♣ Conduct awareness programs for water conservation and sustainable development.
- ♣ Establish institutional environment policy and set an example for environmental responsibility and practices of resource conservation, recycling and waste management.
- ♣ Involve all stakeholders and encourage involvement of government, founders and industry in supporting interdisciplinary research, education, policy formation, and information exchange in water conservation and sustainable development.
- ♣ Collaborate for interdisciplinary approaches to develop curricula, research initiatives, operations, and outreach activities that support an environmentally sustainable future.
- ♣ Promote 3R education policy (reduces, reuse, and recycle) in campus.
- ♣ Arrange training programmes on water management system and nature conservation.
- Lensure participation of students and teachers in local water issues.
- ♣ Conduct seminars, workshops and exhibitions on water and environmental education.





#### CHAPTER-1 INTRODUCTION

#### **1.1 About University**

Kurukshetra University is one of the premier educational institutions of India. Established in 1956, the University is providing higher education to over half a million students on the campus and its affiliated colleges. By virtue of its commitment to excellence, the University has been awarded A<sup>+</sup> Grade by NAAC and has been placed at 8<sup>th</sup> position amongst the State Universities of the country in Category-I by the Ministry of Human Resource Development (MHRD) and has also been granted academic autonomy. Located on the southern bank of famed Brahm Sarovar, Kurukshetra University has a sprawling campus spread over 473 acres. In addition to providing quality education to the students, Kurukshetra University embodies the values of Yogastha Kuru Karmani, enshrined in Bhagwadgita, which means performing one's activities while remaining steadfast in Yoga.

The University is in the rapid process of transformation to an institution of excellence. In this process, digitalization of the University is being done to develop an efficient and transparent system that would further facilitate a learner's experience in the University.

The University programmes combine the enduring value of a liberal arts education with the skills and experience offered by professional departments. The University offers 175 courses on the campus in 47 Departments/institutes through a highly qualified faculty. The University also plays an important role in providing higher education to the youth of the State through its 282 affiliated colleges and institutes in seven districts of Haryana.

The campus of the University has often been rated as one of the most beautiful campuses in India. It resembles a large, self-contained educational village with lecture theatres, smart class rooms, Wi-Fi campus, 24 hrs library facility, laboratories, on campus hostel accommodation, cafeterias, canteens, market, swimming pool, gymnasia, bank, ATM's, post office and world-class sports facilities. The most remarkable feature of the campus is a seamless interconnection of nature and the built environment. There are a number of lush green gardens, water fountains and sidewalks which provide an ideal environment on the campus for study and leisure.







Figure:- Image of KUK from Google map

#### **University Faculties:-**

- ♣ Faculty of Arts & Languages
- ♣ Faculty of Social Sciences
- ♣ Faculty of Life Sciences
- ♣ Faculty of Science
- ♣ Faculty of Education
- ♣ Faculty of Indic Studies
- ♣ Faculty of Engg. & Technology
- **♣** Faculty of Law
- **♣** Faculty of Commerce & Management





#### 1.2 About University Campus:

Kurukshetra University, Kurukshetra (KUK) is a University established on 11 January 1956 in Kurukshetra, in the Indian state of Haryana, 160 kilometres (99 mi) from the capital, Delhi. It is a member of Association of Commonwealth Universities.

Kurukshetra University spread over 473 acres (1.91 km²), the KUK campus is located on the western bank of Brahma Sarovar in the Hindu holy city of Kurukshetra

#### Colleges and schools in Kurukshetra University

The University has these colleges, schools and institutes

- University Senior Secondary Model School
- ♣ Institute of Integrated and Honours Studies (Formerly University College)
- ♣ University College of Education
- **♣** Institute of Environmental studies
- Department of Law
- **♣** Department of Geophysics
- University School of Management
- University Institute of Engineering and Technology
- **♣** Institute of Management Studies
- ♣ Institute of Mass Communication & Media Technology





Table 1.1: -Details of Building and department in KUK University: -

Sr. No	Administration
1	Office of Vice Chancellor
2	Office of the Registrar
	Important Functionaries
1	Deans
2	Other Offices
3	Chairpersons/Directors
4	Teaching Department
5	Deputy Registrar/Assistant Registrar/Admin. Officers
6	Office of the Dean Academic Affairs
7	Office of Dean Faculty of Science
8	Office of Dean Student Welfare
	Accounts Departments
1	Main Accounts
2	Bill Section
3	Cash and Fee
4	Income
5	Pension Cell
6	Fee Section
7	PF Section
8	Planning Section
9	Cheque Section
10	Audit Internal
11	State Govt (Audit)
12	Establishment Branch (Teaching)
13	Establishment Branch (Non-Teaching)
14	Academic Branch
15	Syllabus Branch
16	Scholarship Branch
17	Registration Branch
18	PH.D Branch
19	General Branch
20	Colleges Branch
	Examination-1
1	Confidential Cell
2	Conduct Branch (Theory)
3	Conduct Branch (Practical)
4	UMC
5	PH.D
6	Re-Evaluation Branch
7	Secrecy Branch
	Examination-2





1	Examination Branch -2
2	Result Branch-1
3	Result Branch-2
4	Computer Lab
5	Result Branch -3
6	Certification Section
7	Construction Branch
	University Teaching Department/Institute
1	AIH-Culture & Archaeology
2	Bio-Chemistry
3	Bio-Technology
4	Botany
5	Chemistry
6	Commerce
7	Computer Science & Application
8	Economics
9	Education
10	Electronic Science
11	English
12	Fine Arts
13	Foreign languages
14	Geography
15	Geology
16	Geo-Physics
17	Hindi
18	History
19	Home Science
20	Instrumentation
21	Inst.of Environmental Studies
22	Inst. Of Pharmaceutical Sciences
23	Inst.of Law
24	Inst. of Mass communication & Media Technology
25	Inst. of Management Studies
26	Law
27	Library and Information Science
28	Mathematics
29	Micro-Biology
30	Music and Dance
31	Punjabi
32	Philosophy
33	Physical Education
34	Psychology
35	Public Administration
36	Physics
37	Political Science





20	
38	Sanskrit, Pali & Prakriti
39	Sanskrit and Indological Studies
40	Social Work
41	Sociology
42	Statistics and O R
43	Tourism and Hotel Management
44	University Institute of Eng. & Technology
45	University School of Management
46	Women's Study Research Centre
47	UIET Building
48	Public Administration department
49	ION Beam Centre
50	Zoology
	Institute/College/School at the Campus
1	University College (Institute of Integrated Honors Studies)
2	University College of Education (Institute of Teacher Training & Research)
3	University Sr. Sec. Model School
	Directorates Centres
1	Directorate of Distance Education
2	UGC Human Resource Development Centre
3	Computer Centre
4	Mahatma Gandhi Coaching Centre
5	Dr. B. R Ambedkar Studies Centre
6	Gulzari Lal Nanda Centre of Ethics & Philosophy
7	University Health Centre
	Hostels Office
1	Chief Warden (Boys)
2	Chief Warden (Girls)
	Hostel (Boys)
1	Ambedkar Bhawan
2	Arjun Bhawan
3	Ch. Devi Lal Bhawan
4	Harsh Bhawan
5	Narhari Bhawan
6	Pratap Bhawan
7	Ch. Ranbir Singh Bhawan
8	Shaheed Bhagat Singh Bhawan
9	Swami Vivekanand Bhawan
10	Subhash Bhawan
11	Bheema Bhawan
12	Tagore Bhawan
13	International Bhawan





Hostel (Girls)									
1	Ahilaya Bhawan								
2	Bharti Bhawan								
3	Devyani Bhawan								
4	Ganga Bhawan								
5	Gargi Bhawan								
6	Kasturba Bhawan								
7	Kalpana Chawla Bhawan								
8	Laxmi Bai Bhawan								
9	Meera Bhawan								
10	Sarswati Bhawan								
11	Subhadra Bhawan								
12	Uttara Bhawan								
13	Yamuna Bhawan								
	Services / Facilities								
1	Jawahar Lal Nehru Library								
2	Public Relations Office								
3	Dharohar Haryana Museum								
4	Youth & Cultural Affairs								
5	University Printing & Publications Bureau								
6	NSS								
7	NCC								
8	Directorate of Sports								
9	Land & Farming Department								
10	Horticulture Department								
11	Security Department								
12	Employment Bureau (University)								
	Associations								
1	Kurukshetra University Teaching 'Association (KUTA)								
2	Kurukshetra University Nonteaching Employees Association (KUNTEA)								
3	Kurukshetra University Kurukshetra Alumni Association (KUKAA)								
	General								
1	Post Office and Banks on Campus								
2	NIT Kurukshetra								
3	District Administration								
3	University Health Centre and Doctors (On University Penal of Kurukshetra)								
4	Press Correspondents/Journalists/Print Media								
5	Electronic Media								
6	Sport Education Building								
7	Non-Teaching club								
8	Senior Model School								
9	Construction Branch								
10	University Market								





### 1.3 Green Monitoring Committee: -

INTERNAL QUALITY ASSURANCE CELL KURUKSHETRA UNIVERSITY KURUKSHETRA (Established by the State Legislature Act XII of 1956) ('A+' Grade, NAAC Accredited) Dated: 06.04.2021 E-mail:- head.iqac@kuk.ac.in NOTIFICATION The IQAC Committee in its meeting held under the Chairmanship of the Vice-Chancellor on March 15, 2021 has constituted a Committee of following members to frame the modalities for Green audit, Energy audit, and Environment audit of the University and also suggest the ways for environmental promotional activities beyond the campus, sensor based energy conservation. (i) Prof. Smita Chaudhary (Convener), Institute of Environmental Studies. (ii) Dr. Ram Avtar, Electrical Department, UIET (iii) Dr. Pooja Arora, Institute of Environmental Studies (iv) Dr. Dipti Grover, Institute of Environmental Studies (v) XEN/SDO (Electrical), Construction Branch Director, IOAC Endst. No. IQAC/21/ 932 (1-10) Dated: 06.04.2021 Copy to:a) All the members of the Committee for necessary actions. b) Chairpersons/Directors of the concerned Departments/Institutes. c) P.A. to the Vice-Chancellor, KUK (for kind information of the Vice-Chancellor) d) PS to the Registrar, KUK (for kind information of the Registrar) Internal Quality Assurance Kurukshetra University, Kuruk

#### 1. 4 Energy Audit Team

The study team constituted of the following senior technical executives from **Empirical Exergy Private Limited**,

- **Er.Rajesh Kumar Singadiya** [Director & Accredited Energy Auditor AEA-0284]
- **Mr. Rakesh Pathak**, [Director]
- **Dr. Suresh Soni** [Reviewer]
- **♣ Mr. Sachin Kumawat** [Project Engineer]
- **Mrs. Laxmi Raikwar Singadiya**, [Energy Engineer]
- Mr. Ajay Nahra, [Site Engineer]
- Mr. Akash Kumar, [Site Engineer]
- **Mr. Purushottam Bhade**, [Site Engineer]





#### 1.5 About Environment Auditing

Environment audits can be a highly valuable tool for institute in a wide range of ways to improve their energy, environment and economic performance while reducing wastages and operating costs. Water audits provide a basis for calculating the economic benefits of water conservation projects by establishing the current rates of water use and their associated cost.

#### 1.6 Objectives of Environment audit

The general objective of environment audit is to prepare a baseline report on water conservation measures to mitigate consumption, improve quality and sustainable practices and waste management.

#### The specific objectives are:

- **↓** To monitor the water consumption and water conservation practices.
- ♣ To assess the quantity of water, usage, quantity of waste water generation and their reduction within the University.
- **♣** To assess the waste management practices
- **♣** To assess the rain water harvesting systems in the campus

#### 1.7 Target Areas of Environment audit

This indicator addresses water sources, water consumption, irrigation, storm water, appliances and fixtures aquifer depletion and water contamination are taking place at unprecedented rates. It is therefore essential that any environmentally responsible institution should examine its water use practices and waste management practices.





#### 1.8 Methodology followed for conducting Environment audit

#### **Step 1: Walk through survey**

- ♣ Understanding of existing water sourcing, storage and distribution facility.
- Assessing the water demand and water consumption areas/processes.
- Preparation of detailed water circuit diagram.
- **↓** Identifying the units for rain water harvesting

#### **Step 2: Secondary Data Collection**

- ♣ Analyse historic water use and wastewater generation
- Field measurements for estimating current water use
- Metered & unmetered supplies.
- ♣ Understanding of "base" flow and usage trend at site
- **♣** Past water bills
- **♣** Wastewater treatment scheme & costs etc.

#### **Step 3: Site Environment Audit Planning (based on site operations and practices)**

- ♣ Preparation of water flow diagram to quantify water use at various locations
- **♣** Wastewater flow measurement and sampling plan

#### Step 4: Conduction of Detailed Environment Audit & Measurements

- Conduction of field measurements to quantify water/wastewater streams
- **♣** Power measurement of pumps/motors
- ♣ Preparation of water balance diagram
- **Lestablishing water consumption pattern**
- Detection of potential leaks & water losses in the system
- Assessment of productive and unproductive usage of water
- ♣ Determine key opportunities for water consumption reduction, reuse & recycle.
- Determine key opportunities for waste reduction and management
- Identify RWH systems in the campus

#### **Step 5: Preparation of Environment Audit Report**

- Locumentation of collected & analysed water balancing and measurement details
- ♣ Projects and procedures to maximize water savings and minimize water losses.
- Opportunities for water conservation based on reduce/recycle/reuse and recharge options (RWH systems)
- ♣ Opportunities for waste reduction, recycling and management
- Opportunities for RWH systems and water





### CHAPTER- 2 WATER CONSUMPTION AND WASTE WATER SOURCES

#### 2.1 Details of Source of Fresh Water and Use Areas:

The main source of freshwater is Borewell for the University. The freshwater is mainly used for drinking, housekeeping, gardening, domestic activity and new construction project. Details of the pumps are given below in table.

Table: 2.1 Details of Fresh water sources and Supply pumps

Sr. No	Source of Water	Location	Type of Pumps	Rated (HP)
1	Borewell-02	Rose Garden	Submersible	30 HP
2	Borewell-03	Near Ambedkar Bhavan Hostel	Submersible	30 HP
3	Borewell-04	University 3 <sup>rd.</sup> Gate	Submersible	30 HP
4	Borewell-05	Opposite of University Sr. Sec.  Model School	Submersible	30 HP
5	Borewell-06	Near Girl Hostel No-13 Part-E	Submersible	30 HP
6	Borewell-07	Near Harsh Bhavan Boys Hostel	Submersible	30 HP
7	Borewell-08	Near Shooting range KUK	Submersible	30 HP







#### 2.2 Water Accounting & Metering system:

It was observed that University has installed mechanical type water flow meters on Borewells to quantify per day ground water extraction from different sources. But Audit team observed that water meter was not in working condition.





Figure: -2.1 fresh water supply from Borewell for University campus





#### 2.3 Water Storage Capacity in University Campus: -

There are different types of tank available in University for water storage like Underground RCC tank, Overhead RCC tank and PVC tank (Sintex)

Table 2.2: - Water Storage tank Phase-01 in KUK.

Sr.	Locations	10000	2000	1000	500	300	200
No.		(Litter)	(Litter)	(Litter)	(Litter)	(Litter)	(Litter)
1	TF 1 – 32	0	0	7	2	0	0
2	BTH 1 – 26	0	0	3	2	0	1
3	D 1 -50	0	0	0	50	0	0
4	D 153 -164	0	0	10	0	0	0
5	MTH 1 – 24	0	0	5	0	0	2
6	F 1 – 5	0	0	0	4	0	4
7	F-5	0	0	0	1	0	0
8	E-1 to E-34	0	0	0	34	0	0
9	Teacher club	0	0	0	2	0	0
10	Health Centre	0	0	0	5	1	0
11	R.K Sadan	0	0	0	4	0	0
12	BEd College or Parking	0	0	3	1	0	1
13	UIET Workshop	0	3	2	0	0	0
14	UIET Department	0	2	1	1	0	0
15	Community Centre	0	3	3	1	0	0
16	Old Law	0	1	3	1	0	0
17	Computer Centre	0	0	3	3	0	0
18	Admin Block	0	0	2	0	0	0
19	R-I	0	0	4	0	0	0
20	R-II	0	4	0	0	0	0
21	R-III	0	0	2	0	0	0
22	Hotel Management	0	2	1	1	0	0
23	MBA	0	0	1	3	0	0
24	Ambedkar Bhawan	0	0	0	1	0	0
25	Fine Art	0	0	4	2	1	0
26	VC Office & Registrar Office	0	0	1	2	3	0
27	Commerce and University School of Management	0	0	0	2	1	0
28	Auditorium Hall	0	2	0	2	1	0
29	Darohar Old	0	0	3	2	0	0
30	Darohar New	0	0	0	3	0	0
31	Main Store	0	0	0	0	1	0
32	Law Canteen	0	0	0	0	1	0
33	Pension Cell	0	0	2	0	0	0
34	Centre for advanced research In Earthquake Studies	0	0	0	1	0	0





35	Faculty G. House	0	2	4	0	0	0
36	Inter G. House	0	2	16	0	0	0
37	VC Residence	0	0	0	5	2	0
38	Botany Department	0	0	2	0	0	0
39	Electronic Department	0	0	1	1	0	0
40	Institute of Environmental Studies	0	4	0	0	0	0
41	IIE	0	0	2	0	0	0
42	IOM	0	0	1	0	0	0
43	Botany Park	0	0	1	1	0	0
44	Animal House	0	0	1	1	0	0
45	Coffee House	0	0	3	0	0	0
46	Old Art Faculty	0	0	4	0	0	0
47	New Art Faculty	0	4	0	0	0	0
48	Old Library	0	0	4	0	0	0
49	New Library	0	0	1	1	0	0
50	Rose Garden	0	0	0	1	0	0
51	B - Pharmacy	0	4	0	0	0	0
52	New Chemistry Department	1	0	0	0	0	0
53	Old Chemistry Department	1	0	0	0	0	0
54	USIC Department	0	0	1	1	0	0
55	Physics Department	1	0	0	0	0	0
56	Zoology Department	1	0	0	0	0	0
57	Mathematics Department	0	2	0	0	0	0
58	Central Canteen	0	0	1	0	0	0
59	Student Home	0	0	0	1	0	0
60	University College	0	4	1	2	0	0
61	University College Canteen	0	0	0	2	0	0
	Total	4	39	103	146	11	8





Water Storage Tank Phase -02 in KUK.

Sr. No.	Locations	2000 Litter	1000 Litter	500 Litter	300 Litter	200 Litter	30000 Litter
1	Pratap Bhawan	1	8	0	6	0	0
2	Narhari Bhawan	1	9	0	6	0	0
3	Arjun Bhawan	0	5	0	1	0	0
4	Tagore Bhawan	0	2	2	2	0	0
5	Bhagat Singh	0	8	0	0	0	0
6	Bhima Bhawan	0	3	0	0	0	0
7	Devi Lal Bhawan	0	4	0	0	0	0
8	Ambedkar Bhawan	0	2	1	0	0	0
9	Ranbir Bhawan	0	4	3	0	0	0
10	Harsh Bhawan	0	5	1	2	0	0
11	Vivekananda Bhawan	0	4	0	0	0	0
12	Subhash Bhawan	1	6	6	0	0	0
13	Saraswathi Bhawan	7	3	2	0	1	0
14	Meera Bhawan	10	0	1	0	1	0
15	Kasturba Bhawan	7	6	1	0	1	0
16	Bharti Bhawan	7	2	0	0	1	0
17	Gargi Bhawan	9	1	1	0	1	0
18	Subhadra Bhawan	1	3	1	0	1	0
19	Aahilya Bhawan	0	7	0	1	0	0
20	Ganga Bhawan	2	0	1	0	0	0
21	Uttara Bhawan	0	9	1	0	0	0
22	Devyani Bhawan	2	5	2	0	0	0
23	Kalpana Chawla Bhawan	0	5	1	0	0	0
24	Laxmi Bhawan	0	2	4	0	0	0
25	Yamuna Bhawan	0	2	1	0	0	1
26	D-129 to D - 152	0	0	24	0	0	0
27	E-62 to E -73	0	0	15	0	0	0
28	E-46 to E-61	0	0	33	0	0	0
29	E-35 to E-37	0	0	6	0	0	0
30	E-38 to E-45	0	0	16	0	0	0
31	D-51 to D-70	0	0	20	0	0	0
32	D-72 to D-100	0	0	29	0	0	0
33	New Gym	0	0	3	0	0	0
34	Old Gym	0	0	2	0	0	0
35	Sports Ground	0	0	2	1	0	0
36	DDC	0	2	0	1	2	0
37	New DDC	0	4	1	0	0	0
38	Power House	0	0	1	0	0	0
39	Electricity Board	0	0	1	0	0	0
40	Mass Communication	0	0	3	0	0	0





41	Girls Hostel Barrier	0	0	1	0	0	0
42	KR-1 to KR-24	0	0	6	0	0	0
43	KR General Washroom	0	0	1	0	0	0
44	KR-25 to KR-40	0	0	4	0	0	0
45	C-1 to C-20	0	0	20	0	0	0
46	C-21 to C-40	0	0	20	0	0	0
47	C-41 to C-60	0	0	20	0	0	0
48	C-61 to C-100	0	0	40	0	0	0
49	C-105 to C-140	0	0	35	0	0	0
50	BC-1 to BC-4	0	0	4	0	0	0
51	A-1 to A-6	0	0	6	0	0	0
52	H-1 to H-196	0	0	30	50	0	0
53	University Market	0	0	1	0	0	0
54	SBI Bank	0	0	1	0	0	0
55	Post office	0	0	1	0	0	0
56	Non-Teaching Club	0	0	2	0	0	0
	Total	48	111	377	70	8	1





### 2.4 Water use areas in University Campus: -

Water is preliminary used for drinking, domestic, gardening and clinical activity. Audit team visited various departments and buildings to determine appliances. The details of washrooms, toilets and taps are given in table.

Table: 2.3 Details of various water Taps in Girls Hostel

Sr. No	Area Name	owc	EWC	Geyser	Pillar Cock	Bib Cock	Shower	Concealed / stop cock	Hydrants	Urinal	Total
1	G. Hostel No. 1	24	12	17	37	125	34	34	3	0	286
2	G. Hostel No. 2	24	11	17	37	119	35	35	3	0	281
3	G. Hostel No. 3	24	12	17	37	119	35	35	4	0	283
4	G. Hostel No. 4	24	11	17	37	119	35	35	3	0	281
5	G. Hostel No. 5	23	12	17	28	146	35	35	3	0	299
6	G. Hostel No. 6	23	3	14	28	90	25	25	4	0	212
7	G. Hostel No. 7	1	24	23	55	88	25	48	1	0	265
8	G. Hostel No. 8	12	2	5	16	51	13	13	2	0	114
9	G. Hostel No. 9	46	23	26	50	245	70	70	4	0	534
10	G. Hostel No. 10	31	12	20	32	157	46	46	2	0	346
11	G. Hostel No. 11	20	14	11	44	115	33	33	2	0	272
12	G. Hostel No. 12	23	12	11	44	115	40	40	3	0	288
13	G. Hostel No. 13	15	28	25	54	173	48	48	1	0	392





14	G. Hostel Market (Complex)	2	0	0	3	5	0	0	0	0	10
15	G. Hostel Canteen	1	0	1	1	5	0	0	0	0	8
16	Guest House	0	2	2	2	8	2	2	0	0	18
17	Hostel Dispensary & Gym	3	2	0	3	9	0	0	0	0	17
18	Servant Qtr. (G.H)	7	1	0	9	26	0	0	0	0	43
20	Security Room - I, II (G.H)	2	0	0	3	8	0	0	0	2	15
	Total	305	181	223	520	1723	476	499	35	2	3964





Table 2.4: Details of Various water Taps in Boys Hostel

Sr. No	Area Name	owc	EWC	Geyser	Pillar Cock	Bib Cock	Shower	Hydrants	Urinal	Total
1	Narhari Hostel	32	30	17	63	188	60	4	46	440
2	Pratap Hostel	32	30	17	65	192	60	2	46	444
3	Bhima Hostel	32	2	17	30	103	1	1	13	199
4	Subhash Bhawan	0	17	16	38	55	2	16	0	144
5	Ambedkar Bhawan	8	5	6	1	46	1	0	8	75
6	Tagore Bhawan	25	3	12	23	93	4	27	22	209
7	Ranbir Bhawan	29	15	14	35	134	2	38	19	286
8	Devi Lal Bhawan	15	13	7	24	99	0	3	16	177
9	Police Post	0	1	0	1	1	0	0	0	3
10	Bhagat Singh Bhawan	33	20	17	74	174	1	1	33	353
11	Power House No3	1	0	0	1	2	0	0	0	4
12	Vivekanand Bhawan	10	5	7	15	53	0	3	12	105
13	Harsh Bhawan	32	16	17	35	162	46	4	36	348
14	Arjun Bhawan	20	1	11	26	74	21	3	21	177
15	New Servant Qtr (Behind Arjun Bhawan)	6	0	0	2	14	0	0	0	22
16	Old Servant Qtr (i)	1	0	0	1	3	0	0	0	5
17	Old Servant Qtr (ii)	5	0	0	0	7	0	0	0	12
18	New Block (ii)	8	0	0	2	16	0	0	0	26
19	Extra 14,16, 18 Qtr.	9	1	-	12	40	0	0	0	62
20	Servant Qtr. (Near Tagore Bhawan)	10	0	0	2	52	0	0	0	64
	Total	308	159	158	450	1508	198	102	272	3155





Table 2.5 :- Details of Various water Taps in Residential Area

Sr. No.	Particular	Pillar Cock	OWC/EWC	Taps	JET	Water Cooler	Urinal	Stop Cock / Consoled	Shower	Hydrants	Total
1	Rose Garden	3	6	18	3	1	2	0	0	8	41
2	M.T.H (1 TO 24)	48	26	73	26	0	0	24	24	1	222
3	B.T.H (1 TO 24)	24	24	120	24	0	0	24	24	1	241
4	B.T.H ( 25 TO 26)	4	3	13	1	0	0	2	2	0	25
5	Teacher Flat (1 to 32)	64	64	265	34	0	0	64	64	1	556
6	D-1 To D-50	100	100	600	50	0	0	100	100	0	1050
7	D 153 To D 164 Stilt Floor	53	77	188	41	0	0	41	41	0	441
8	F-Type (1 to 05)	20	15	40	15	0	0	15	15	0	120
9	F-Type (1 to 05) Waiter Room	-	5	7	1	0	0	0	0	0	12
10	E - 1 to E- 30, E-33 & E-34	147	127	535	86	0	0	79	79	32	1085
11	Old Rest House	6	6	21	6	5	0	15	5	0	64
12	Rest House	51	69	175	62	1	7	192	52	5	614
13	V.C. Residence	6	6	19	5	2	2	3	3	4	50
	Total	526	528	2074	353	9	11	559	409	52	4521





Sr. No	Area Name	owc	EWC	Geyser	Pillar Cock	Shower	Stop Cock	Hydrants	Toilet (Bath) W.C. Sink Tap	Total
1	A-1 to a-6	6	0	1	6	6	6	0	24	49
2	BC-1 to BC - 4	2	4	3	4	4	4	0	22	43
3	C-1 to C-20	20	20	40	40	40	40	0	197	397
4	C-21 to C-40	20	20	24	32	32	32	0	116	276
5	C-41 to C-60	20	20	32	40	40	40	0	192	384
6	C-61 to C-80	20	20	32	40	40	40	0	192	384
7	C-81 to C-100	20	20	33	40	40	40	0	193	386
8	C-105 to C-124	20	20	36	40	40	40	0	197	393
9	C-125 to C-140	16	16	23	32	32	32	0	174	325
10	CF -1 to CF-4 (Near NCC)	4	0	2	4	4	4	0	12	30
11	CF -1 to CF-4 (Near Residence Quarters)	4	0	2	4	4	4	0	12	30
12	CI-1 to CI -48	48	48	96	100	96	96	0	436	920
13	H-1 to H-15	12	3	0	15	15	15	0	75	135
14	D-51 to D-62	9	27	24	26	24	24	0	158	292
15	D-63 to D-70	8	16	16	17	16	16	0	96	185
16	D-72 to D-78	4	10	14	21	14	14	0	72	149
17	D-79 to D-94	12	20	20	51	32	35	0	152	322
18	D-95 to D-100	0	12	6	18	12	14	0	60	122
19	D-101 to D-106	2	10	12	20	12	14	0	74	144
20	D-101A to D-101C	1	3	6	12	6	12	1	30	71
21	D-129 to D-152	24	24	48	96	48	96	0	290	626
22	D-107 to D-112	4	8	12	30	12	24	0	52	142
23	D-113 to D-118	4	9	12	31	13	25	0	56	150
24	D-119 to D-124	4	8	12	30	12	24	0	53	143





25	D-125 to D-128	3	5	8	20	8	16	0	30	90		
26	E-62 to E-73	2	37	37	76	38	76	0	192	458		
27	E-54 to E-61	0	16	16	40	16	32	0	80	200		
28	E-46 to E-53	2	16	16	43	18	34	0	92	221		
29	E-35 to E-45	11	22	22	56	33	40	0	145	329		
30	E-31 to E-32	4	0	2	4	4	4	0	18	36		
31	H-16 to H-30	13	2	0	15	15	15	0	75	135		
32	H-115 to H-129	12	3	3	15	15	15	0	78	141		
33	H-130 to H-144	15	0	0	15	15	15	0	66	126		
34	H-31 to H-45	15	0	3	15	15	15	0	66	129		
35	H-46 to H-60	15	0	0	15	15	15	0	60	120		
36	H-61 to H-69	9	0	2	9	9	9	0	42	80		
37	H-70 to H-84	15	0	0	15	15	15	0	60	120		
38	H-85 to H-99	15	0	0	15	15	15	0	60	120		
39	H-100 to H-114	15	0	2	15	15	15	0	64	126		
40	H-155 to H-164	7	3	3	10	10	10	0	56	99		
41	H-181 to H-196	15	1	2	16	16	16	0	66	132		
42	H-145 to H-152	8	0	0	8	8	8	0	32	64		
43	H-165 to H-174	10	0	0	10	10	10	0	40	80		
44	K.R -01 to K.R-24	24	0	1	19	3	3	0	96	146		
45	K.R -25 to K.R-28	4	0	1	4	2	2	0	16	29		
46	K.R -29 to K.R-32	4	0	2	4	4	4	0	16	34		
47	H-175, 176, 153,154	4	0	1	4	4	4	0	16	33		
48	K.R-33 to K.R-36	4	0	0	4	2	2	0	16	28		
49	H-177 to H-180	4	0	1	4	1	1	0	20	31		
50	K.R 37 to K.R 40	4	0	0	4	0	0	0	16	24		
51	Common Toilet K.R	6	0	0	2	0	0	0	38	46		
			Total									





Table 2.6 :- Details of Various water Taps in various Buildings

Sr. No.	Area Name	Pillar Cock	OWC/EWC Cistern	Taps	EWC JET	Water Cooler	Urinal	Stop Cock / Consoled	Shower	Total
1	Faculty Lounge	6	6	8	4	0	3	0	0	27
2	V.C office & Registrar office	19	17	42	16	2	5	18	0	119
3	Adm. Block	32	20	38	9	4	12	1	0	115
4	Computer Dept. (DEAN Build.)	40	51	59	27	6	19	12	0	214
5	Conduct Branch	7	5	12	3	1	2	-	0	30
6	Exam -I	25	26	50	13	4	21	12	0	151
7	Exam -2	53	47	58	32	6	35	30	0	261
8	Exam-3	14	18	20	8	1	7	5	0	73
9	Ambedkar Bhawan	7	7	15	5	1	4	0	0	39
10	Social Work Dept.	14	14	18	7	1	11	0	0	65
11	Tourism& Hotel Management	24	22	46	14	3	15	0	0	124
12	Auditorium Hall	33	24	44	12	2	22	0	0	137
13	Dharoher I & II	15	24	26	18	2	5	0	0	90
14	Museum 1857	9	9	11	9	-	8	0	10	56
15	Fine Art	18	27	47	23	2	17	12	0	146
16	Commerce & Uni. School of Mgmt.	23	66	31	6	8	11	0	0	145
17	Pub. Toilet	3	5	6	0	0	2	2	0	18
18	2nd Gate Toilet	0	0	4	0	2	0	0	0	6
19	Main Library	22	23	31	9	3	23	0	0	111
20	Jublee Hall	31	30	44	20	5	20	0	0	150





21	New Art Faculty	27	15	24	11	3	15	0	0	95
22	Old Art Faculty Dept.	27	30	47	12	4	20	0	0	140
23	Coffee House	4	3	10	2	2	0	0	0	21
24	Chief Warden Off. Boys & Girls	2	2	4	2	0	0	0	0	10
25	Modern Chem. Lab	27	21	227	12	1	27	101	0	416
26	Chemistry Dept.	9	7	162	6	2	6	152	0	344
27	USIC Dept.	12	9	23	5	2	8	-	0	59
28	Physics Dept.	10	9	83	4	3	12	78	0	199
29	Zoology Forensic Science lab	5	5	18	5	1	2	0	0	36
30	Zoology / Maths / Geography Dept.	13	16	107	9	3	11	83	0	242
31	B. Pharmacy	17	17	191	10	5	8	456	0	704
32	IIE	13	11	13	7	1	4	0	0	49
33	ION Beam Centre	4	5	10	4	2	2	0	0	27
34	Public Admin / Eng. Statistics	27	27	41	15	4	20	0	0	134
35	Geophy / Phycology / Women Study	15	15	27	9	3	9	0	0	78
36	Dept. Geo Physics & Music	40	32	47	16	5	12	0	0	152
37	IIHS	32	16	64	5	2	14	63	0	196
38	Central Canteen	2	2	2	1	1	1	0	0	9
39	Student Home & Empl. Bureau	2	4	8	3	1	1	0	0	19





40	B Ed College	13	12	21	0	2	10	1	1	60
41	R.K. Sadan	7	7	7	5	0	2	0	0	28
42	3rd Gate Toilet	2	4	8	2	1	3	0	0	20
43	Parking Toilet	4	5	7	2	1	4	2	0	25
44	UIET Workshop	36	29	66	21	4	16	0	0	172
45	UIET Dept.	40	73	169	35	9	38	91	0	455
46	Community Centre	21	17	52	14	1	8	18	11	142
47	Institute of Law	26	46	50	33	4	9	27	0	195
48	Botany Dept. & Home Science Dept	13	13	160	7	2	11	104	0	310
49	Elect. Sci. / Geology Dept.	13	13	85	7	3	9	66	0	196
50	Animal House	5	5	39	2	1	0	33	0	85
51	Bio Chem./ Biotech / Micro Biology Dept	13	13	94	7	3	10	81	0	221
52	Institute of Environmental Studies	32	33	33	23	3	15	0	0	139
53	University Health Centre	19	13	35	7	1	1	2	0	78
54	Teacher Club	5	4	13	3	1	2	2	2	32
	Total	932	956	2557	524	134	552	1451	24	7130





Sr. no	Area Name	OWC	EWC	Geyser	Pillar Cock	Bib cock	Shower	Hydrants	Urinal	Total
1	New Multimedia (Block-C)	4	4	0	8	8	0	0	4	28
2	B. Tech (Block-B)	1	2	0	3	7	0	0	1	14
3	Old D.C.C.	6	8	0	15	25	0	3	8	65
4	Sports Ground	4	3	0	7	16	3	9	4	46
5	New Mass Comm.	4	8	0	11	14	0	0	6	43
6	University Model Sr. Sec. School	11	6	0	19	65	0	7	6	114
7	New D.C.C	10	12	0	22	37	0	0	12	93
8	Mass Studio	2	2	0	2	6	0	0	0	12
9	Farm House	1	1	2	2	14	2	1	0	23
10	Dept. of Phy Edu.	8	7	2	17	41	4	0	8	87
11	Non-Teaching	4	-	-	3	10	1	0	0	18
12	Post office	1	2	1	4	10	1	0	2	21
13	Construction Branch	2	5	0	5	12	0	1	5	30
14	Press	2	3	0	7	11	0	0	5	28
15	Shop No. 1 to 9, 10,11,55	3	7	0	6	30	0	0	2	48
16	SBI Bank	0	2	0	1	5	0	0	2	10
17	Shop No. 30 to 43	0	0	0	0	2	0	0	0	2
18	Market Common Toilet/ Sink	3	1	0	3	9	0	0	3	19
19	Shop No. 44 to 53	0	0	0	1	0	0	0	0	1
20	Shop No. 12 to 29	0	0	0	1	0	0	0	0	1
21	Power House No. (1 & 4)	1	1	0	1	9	0	1	1	14
22	T. W No. 3,5,6,7,8	3	0	0	4	22	1	2	0	32
	Total									





#### 2.5 Fresh Water uses for Gardening:

An irrigation sprinkler (also known as a water sprinkler or simply a sprinkler) is a device used to irrigate (water) agricultural crops, lawns, landscapes, golf courses, and other areas. They are also used for cooling and for the control of airborne dust. KUK is using Water Sprinkler system for garden and lawns. **It is appreciable.** 



Figure: - KUK using Sprinkler system for garden and lawn





The one of major contribution from fresh water consumption is watering for plants and garden in University campus. There is good potential for water saving by adopting "Automatic Watering 360 adjustable misting nozzle irrigation Dripper's system" for plants. Adjustable drip irrigation tools provide different amounts of water depending on the water requirements of different plants. The drip speed can be set accordingly for indoor and outdoor plants.







Fig: 2.5 Proposed Drip irrigation Technology for Plants area.





#### 2.6 Waste Water Generation sources: -

At present, waste water generated from various departments, canteen, Mess, hostels and other activities like washrooms, handwash and washing of medical equipment's and RO rejected etc is treated in STP plant in University.

Table: - 2.7 Waste water generation area in University campus

Sr. No	Key Water Usage Section	Type of water used (raw, treated etc.)	Water Consuming activities
1	Admin Block	Fresh Water	Drinking and other uses
2	Hostels	Fresh Water	Drinking, Food cooking, other Uses
3	Institution Buildings	Fresh Water	Drinking and other uses
4	Canteens	Fresh Water	Food cooking, drinking
5	Residential	Fresh Water	Drinking, domestic and other activities

### ♣ Some photographs of waste water generation sources are given





Figure: -2.6 Waste Water Generation sources

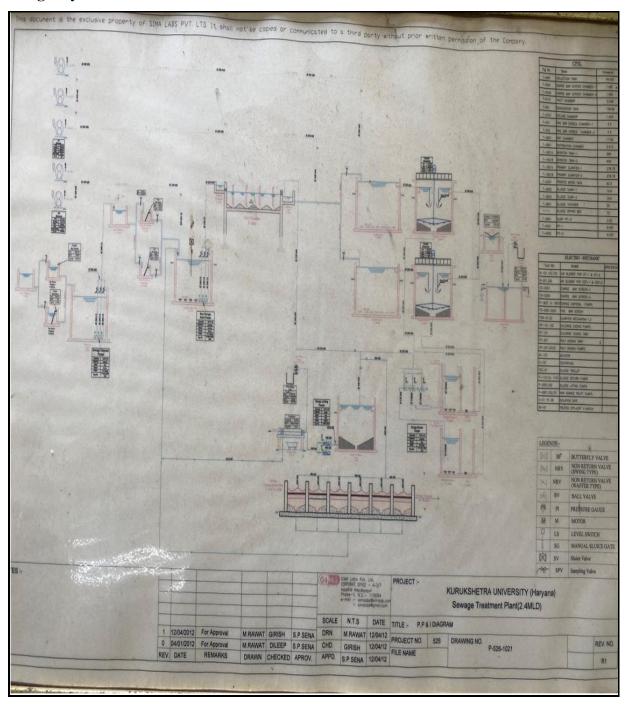




#### 2.7Water Treatment Plant in KUK: -

**♣ STP PLANT:** -University has installed 2.4 MLD Sewage Treatment Plant (STP) for waste water treatment in University premises. Environment audit team visited STP plant area and found out that STP treated water are drained in open area.

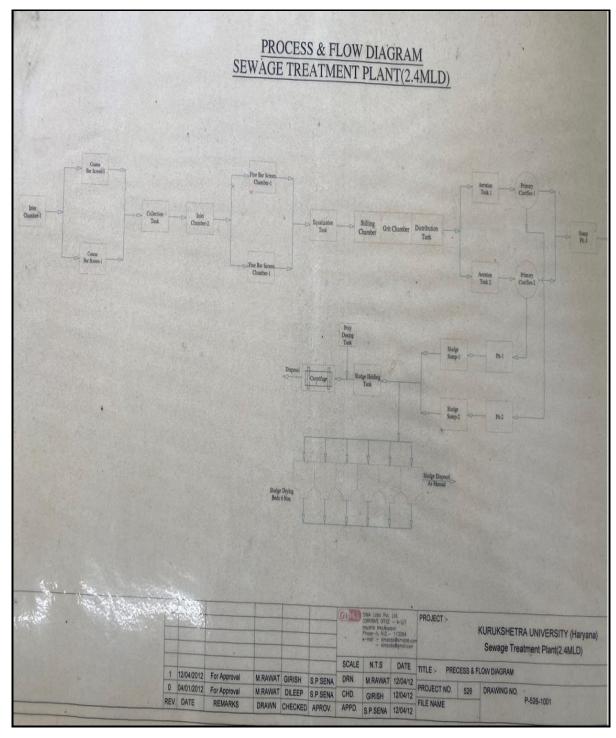
#### Design layout of STP Plant in KUK: -







#### Process and water flow diagram of STP plant: -







#### **Some Photograph of STP Plant: -**











#### ON SITE WATER FLOW MEASURMENT IN STP PLANT: -



Figure :- Water Measurement on University outlet for STP







Figure: - Flow Measurement on STP primary Pump

#### **♣** PRPOSED ETP PLANT: -

University has installed one STP plant for waste water treatment. **It isApplicable.** Environment audit team visited in some departments in KUK they usevarious chemicals in Lab for experiment activities.

Table 2.8:- Details of departments that use Chemicals in laboratory: -

Sr. NO	Department Name	Purpose
1	Chemistry Department	Lab Activity
2	Botany Department	Lab Activity
3	Zoology Department	Lab Activity
4	Biochemistry Department	Lab Activity
5	Microbiology Department	Lab Activity
6	Biotechnology department	Lab Activity
7	Institute of Environment Science	Lab Activity
8	Pharmaceutical Science	Lab Activity
9	IIHS KUK	Lab Activity

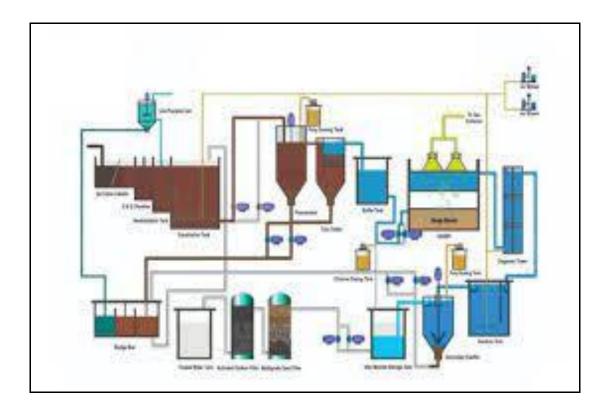
Effluent treatment plant cleans industrial effluents, contaminated water from outlet pipes, reservoir, rivers, lakes etc and reclaims the water resource for using in different purposes. ETPs are mostly installed in industries like textile industry, Medicine manufacturing, leather industry, and chemicals industry.





The principle of operation of ETP is Physico-Chemical treatment followed by Polishing Treatments like –Sand Filtration, Activated Charcoal treatment (Adsorption), Ozonisation (Chemical Oxidation), Ultra Filtration (UF), Reverse Osmosis (RO) and evaporation.

It works by using aeration and flocculation together. As we have seen, wastewater flows into an aeration tank and mixed with water during aeration. Afterward, the wastewater flows into a settling tank or secondary clarifier.









#### Recommendation: -

It is highly recommended to separate all drain water lines that departments use for chemicals discharge from lab activities and proposed to connect to ETP plant for treatment of chemical contaminated water.

#### 2.8:- On site water Measurement in all borewell:-

Table 2.9: - Water Flow measurement on Borewells

Sr. No	Source of Water	Location	Measured Flow (M³/hr)	Rated (HP)	Type of Pumps
1	Borewell-02	Rose Garden	60	30 HP	Submersible
2	Borewell-03	Near Ambedkar Bhavan Hostel	70	30 HP	Submersible
3	Borewell-04	University 3 <sup>rd.</sup> Gate	38	30 HP	Submersible
4	Borewell-05	Opposite of University modal school	112	30 HP	Submersible
5	Borewell-06	Near Girl Hostel No-13 Part-E	71	30 HP	Submersible
6	Borewell-07	Near Harsh Bhavan Boys Hostel	91	30 HP	Submersible
7	Borewell-08	Near Shooting range KUK	58	30 HP	Submersible



















#### 2.9 Water test report of ground water in KUK University: -

Payment Passi 1	EERING DEPARTMENT, HARYANA	
Payment Receipt Of Water Sample Test (Consumer Copy)		
Received from : S.D.O(P.H) K.U.K	Date: 06-Oct-2020 11:35:27 AM	
District : KURUKSHETRA	Lab Name : KURUKSHETRA	
Town : Kurukshetra (Thanesar)	Chemist : SMT. NISHA SAINI	
Ward: 25	Sample Location: UNIVERSITY CAMPUS	
	Address:	
By : Online Payment		
Online Transaction No.: S0ba6b83f4a3a74132a1	GRN No.: 68052791	
Test Name: Physical & Chemical Test		
Test Parameter		
Total Dissolved Solids		
Total Hardness as CaCo3		
Calcium as Ca		
Magnesium as Mg		
Iron as Fe		
Chloride as CI		
Sulphate as So4		
Flouride as F		
Nitrate as No3		
рН		
Alkalinity		
Color		
Odour		
Taste		
Turbidity		
Free Residual Chlorine		
Rs. 250 (In words TWO HUNDRED AND FIFT	Y only)	
RS. 230 (III Words 1996 Protection)		

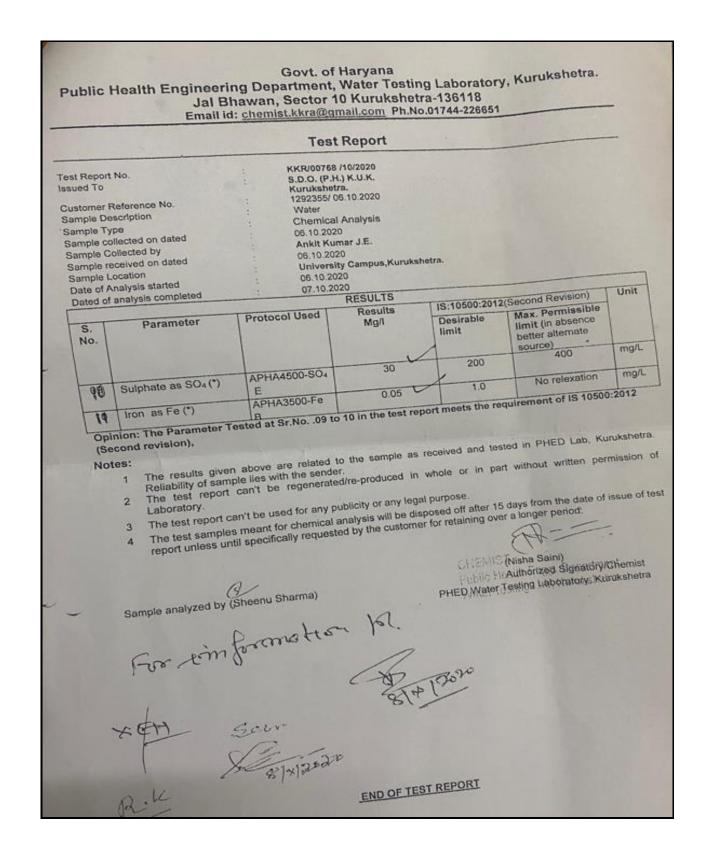




Test Report No. Issued To S.D.O. (P.H.) K.U.K. Customer Reference No. Sample Description Sample Description Sample Description Sample Description Water Chemical Analysis Sample Collected by Ankit Kumar J.E. Sample received on dated O6.10.2020 Sample Location Date of Analysis started O6.10.2020 Sample Location University Campus, Kurukshetra. O6.10.2020  Test University Campus, K	Test Report No. Issued To S.D.O. (P.H.), K.U.K. Customer Reference No. 1292355 06. 10. 2020 Sample Description Sample Description Water Chemical Analysis Sample collected on dated Sample Collected by Ankit Kumar J.E. Sample received on dated Sample Location Date of Analysis started Date Office of Analysis started Date Office of Analysis started Date of Analysis ompleted Protocol Used RESULTS  S. Parameter Protocol Used Results No. Parameter Protocol Used Results No. Parameter Protocol Used Results Solids (TDS) at 180°C  I Total Dissolved Solids (TDS) at 180°C  I Total Hardness IS.3025 (Part 16)1984Reaffirmed) 2017  A Magnesium (as Mg) IS.3025 (Part 46)1994Reaffirmed) 304  A Magnesium (as Mg) IS.3025 (Part 46)1994Reaffirmed) 304  A Magnesium (as Mg) IS.3025 (Part 32)20 BPart B Tiration method Choride (as Cl) Results Results Results Desirable is Man. Permissible limit (in absolved as a control of the control o	Test Report No. Issued To S.D.O. [P.H.], K.U.K. Kurukshetra. 1292355 06. 10.2020 Sample Description Water Chemical Analysis Oat. 10. 2020 Sample Collected on dated 06. 10. 2020 Sample Collected by Anklt Kumar J.E. 06. 10. 2020 Sample Collected by Anklt Kumar J.E. 06. 10. 2020 Sample Location University Campus, Kurukshetra. 06. 10. 2020 Dated of Analysis started 07. 10. 2020 Dated of analysis completed 77. 2020 Dated of analysis completed 77. 2020  Sample Docation 19. 2020 RESULTS  S. Parameter Protocol Used Results Mg/I Desirable Illimit absono before alternate source)  1 Total Dissolved Solids(TD6)at 180°C 18. 3025 (Part 16)1984Reaffirmed)2017 556 500 2000 mg  2 Total Hardness 18. 3025 (Part 21)2009Reaffirmed)2014 320 200 600 mg  3 Calcium (as Ca) IS.3025 (Part 46)1994Reaffirmed)2014 78. 15 75 200 mg  4 Magnesium (as Mg) IS. 3025 (Part 46)1994Reaffirmed)2014 78. 15 75 200 mg  5 Total Alkalinity APHA 2320 B Part B Tiration method 310 200 600 mg  6 Chloride (as Cl) IS. 3025 (Part 39)98Reaffirmed)2014 57. 35 250 1000 mg  8 Nitrate (as NO3) APHA 4500-F Part C ion selective method 1. 67. 35 250 1000 mg  8 Nitrate (as NO3) APHA 4500-F Part C ion selective method 0. 6 45 mg  8 Nitrate (as NO3) APHA 550 No. 7 are leasted to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list of the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list of the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list of the samples are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list of the samples are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list of the samples and the samples are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list of the samples meant for chemical analysis will be disposed of after 15 days from the date pf issue of test report unless until samples meant for chemical analysis will be dis				Test Report				
Customer Reference No. 1292/355 / 06.10 2020 Sample Description Water Date	Customer Reference No.  Sample Description  Sample Description  Sample Collected on dated  Sample Location  University Campus, Kurukshetra.  O6 10 2020  RESULTS  Results  Mg/I  Total Dissolved  Solids(TDS)at 180°C  Total Hardness  Solids(TDS)at 180°C  Total Hardness  Solids(TDS)at 180°C  Total Akailinity  APHA 2320 B Part B Tirraben method  Magnesium (as Mg)  Solids(TOs)at Sample Collected (as Cl)  Fluoride (as Cl)  Solids(Part 16)1984Reaffirmed)2014  APHA 2320 B Part B Tirraben method  Total Akailinity  APHA 2320 B Part B Tirraben method  Cloride (as Cl)  Fluoride (as Cl)  Solids (Part 11)1983Reaffirmed)2014  Notes:  The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample in the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample in the test report unless until 1  The test seport can't be used for any publicity or any legal surpose.  The test report can't be used for any publicity or any legal surpose.  The test report can't be used for any publicity or any legal surpose.  The test report can't be used for any publicity or any legal surpose.  The test report can't be used for any publicity or any legal surpose.  The test report can't be used for any publicity or any legal surpose.  The test report can't be used for any publicity or any legal surpose.  The test report can't be used for any publicity or any legal surpose.	Customer Reference No.  Sample Description  Sample Description  Sample Collected on dated  Sample Location  University Campus, Kurukshetra.  O6 10 2020  RESULTS  Results  Mg/I  Total Dissolved  Solids(TDS)at 180°C  Total Hardness  IS:3025 (Part 16)1984Reaffirmed)2017  Total Hardness  IS:3025 (Part 46)1991Reaffirmed)2014  Magnesium (as Mg)  Magnesium (as Mg)  Sample Location  Mg/I  Magnesium (as Mg)  Sample Location  Mg/I  Mg/I  Sample Collected on dated  Sample Location  Mg/I  S		Test Re	port No.			-		-
Sample received on dated   O6.10.2020   Sample Location   University Campus, Kurukshetra.   Total Campus Started   O6.10.2020   O7.10.2020	Sample Location University Campus, Kurukshetra.  Date of Analysis started 06.10.2020  Date of analysis completed 07.10.2020  RESULTS  S. Parameter Protocol Used Results Mg/I Desirable Illimit Max. Permissible limit (in absence better attenute source)  1 Total Dissolved Solids (TDS) at 180°C IS.3025 (Part 16)1984Reaffirmed)2017 556 500 2000 mm  2 Total Hardness IS.3025 (Part 21)2009Reaffirmed)2014 320 200 600 mm  3 Calcium (as Ca) IS.3025 (Part 40)1991Reaffirmed)2014 76.15 75 200 mm  4 Magnesium (as Mg) IS.3025 (Part 40)1991Reaffirmed)2014 30.43 30 100 mm  5 Total Akalinity APHA 2320 B Part B Titration method 310 200 600 mm  6 Chloride (as Cl) IS.3025 (Part 45)1994Reaffirmed)2014 67.35 250 1000 mm  7 Fluoride (as F) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 mg  8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D Ion Mater electrode 0.6 45 45 mg  Notes:  1 The results given above are related to the sample as received and tested in PHEO Lab, Kurukshetra. Reliability of sample is sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory  The test report can't be used for any publicity or any legal surpose.  3 The test report can't be used for any publicity or any legal surpose.  3 The test samples meant for chamical analysis will be disposed off after 15 days from the date of less until 1 requested by the customer for retaining over a longer period.	Sample Location University Campus, Kurukshetra.  Date of Analysis started 06.10.2020  Date of analysis completed 07.10.2020  S. Parameter Protocol Used Results Mg/I Desirable limit (in absence before retensite source)  1 Total Dissolved Solids (TDS) at 180°C IS.3025 (Part 16)1984Reaffirmed)2017 556 500 2000 mg  2 Total Hardness IS.3025 (Part 21)2009Reaffirmed)2014 320 200 600 mg  3 Calcium (as Ca) IS.3025 (Part 40)1991Reaffirmed)2014 78.15 75 200 mg  4 Magnesium (as Mg) IS.3025 (Part 40)1991Reaffirmed)2014 30.43 30 100 mg  5 Total Akalinity APHA 2320 B Part B Titration method 310 200 600 mg  6 Chloride (as Cl) IS.3025 (Part 40)1991Reaffirmed)2014 67.35 250 1000 mg  7 Fluoride (as Cl) IS.3025 (Part 40)1991Reaffirmed)2014 67.35 250 1000 mg  8 Nitrate (as NO3) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 mg  8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D Ion Mater electrode 0.6 45 45 mg  Ph IS.3025 (Part 11)1983Reaffirmed)2017 7.72 8.5 to 8.5 No relexation pethod (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHEO Lab, Kurukshetra. Reliability of sample list sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory  The test report can't be used for any publicity or any legal surpose.  3 The test report can't be used for any publicity or any legal surpose.  3 The test report can't be used for any publicity or any legal surpose.  3 The test report can't be used for any publicity or any legal surpose.  3 The test report can't be used for any publicity or any legal surpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of less report unless until a requested by the customer for retaining over a longer period.		Issued '	To · .	S.D.O. (P.H.) K.U.K.			Manua No 60	)
Sample received on dated   O6.10.2020   Sample Location   University Campus, Kurukshetra.   Total Campus Started   O6.10.2020   O7.10.2020	Sample Location University Campus, Kurukshetra.  Date of Analysis started 06.10.2020  Date of analysis completed 07.10.2020  RESULTS  S. Parameter Protocol Used Results Mg/I Desirable Illimit Max. Permissible limit (in absence better attenute source)  1 Total Dissolved Solids (TDS) at 180°C IS.3025 (Part 16)1984Reaffirmed)2017 556 500 2000 mm  2 Total Hardness IS.3025 (Part 21)2009Reaffirmed)2014 320 200 600 mm  3 Calcium (as Ca) IS.3025 (Part 40)1991Reaffirmed)2014 76.15 75 200 mm  4 Magnesium (as Mg) IS.3025 (Part 40)1991Reaffirmed)2014 30.43 30 100 mm  5 Total Akalinity APHA 2320 B Part B Titration method 310 200 600 mm  6 Chloride (as Cl) IS.3025 (Part 45)1994Reaffirmed)2014 67.35 250 1000 mm  7 Fluoride (as F) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 mg  8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D Ion Mater electrode 0.6 45 45 mg  Notes:  1 The results given above are related to the sample as received and tested in PHEO Lab, Kurukshetra. Reliability of sample is sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory  The test report can't be used for any publicity or any legal surpose.  3 The test report can't be used for any publicity or any legal surpose.  3 The test samples meant for chamical analysis will be disposed off after 15 days from the date of less until 1 requested by the customer for retaining over a longer period.	Sample Location University Campus, Kurukshetra.  Date of Analysis started 06.10.2020  Date of analysis completed 07.10.2020  S. Parameter Protocol Used Results Mg/I Desirable limit (in absence before retensite source)  1 Total Dissolved Solids (TDS) at 180°C IS.3025 (Part 16)1984Reaffirmed)2017 556 500 2000 mg  2 Total Hardness IS.3025 (Part 21)2009Reaffirmed)2014 320 200 600 mg  3 Calcium (as Ca) IS.3025 (Part 40)1991Reaffirmed)2014 78.15 75 200 mg  4 Magnesium (as Mg) IS.3025 (Part 40)1991Reaffirmed)2014 30.43 30 100 mg  5 Total Akalinity APHA 2320 B Part B Titration method 310 200 600 mg  6 Chloride (as Cl) IS.3025 (Part 40)1991Reaffirmed)2014 67.35 250 1000 mg  7 Fluoride (as Cl) IS.3025 (Part 40)1991Reaffirmed)2014 67.35 250 1000 mg  8 Nitrate (as NO3) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 mg  8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D Ion Mater electrode 0.6 45 45 mg  Ph IS.3025 (Part 11)1983Reaffirmed)2017 7.72 8.5 to 8.5 No relexation pethod (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHEO Lab, Kurukshetra. Reliability of sample list sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory  The test report can't be used for any publicity or any legal surpose.  3 The test report can't be used for any publicity or any legal surpose.  3 The test report can't be used for any publicity or any legal surpose.  3 The test report can't be used for any publicity or any legal surpose.  3 The test report can't be used for any publicity or any legal surpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of less report unless until a requested by the customer for retaining over a longer period.		Custom	er Reference No.				Memo No.	119
Sample Location	Sample Location Date of Analysis started Oated of analysis completed O7.10.2020  RESULTS  Results No. Parameter Protocol Used RESULTS  Results No. Parameter Protocol Used RESULTS  Results No. RESULTS  Results No. Results N	Sample Location Date of Analysis started Oated of analysis completed O7.10.2020  RESULTS  Res							Date//	100
Sample Location University Campus, Kurukshetra.  Date of Analysis started 06.10.2020  Dated of analysis completed 07.10.2020  RESULTS  S. Parameter Protocol Used Results Mg/I S:10500:2012(Second Revision)  Total Dissolved IS:3025 (Part 16)1984Reaffirmed)2017 556 500 2000 r  2 Total Hardness IS:3025 (Part 21)2009Reaffirmed)2014 320 200 600 r  3 Calcium (as Ca) IS:3025 (Part 40)1991Reaffirmed)2014 78.15 75 200 r  4 Magnesium (as Mg) IS:3025 (Part 8)1994Reaffirmed)2014 30.43 30 100 r  5 Total Alkalinity APHA 2320 B Part B Tiration method 310 200 600 f  6 Chloride (as Cl) IS:3025 (Part 40)1998Reaffirmed)2014 67.35 250 1000 r  7 Fluonde (as F) APHA 4500-F Part C ion selective method Less than 1 1.0 1.5 n  8 Nitrate (as NO3) APHA4500-No; Part D ion Mater electrode 0.5 45 45 m  Notes: The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra: Reliability of sample sender.	Sample Location Date of Analysis started Date of Analysis started Date of Analysis completed  Of 10 2020  Dated of analysis completed  Of 10 2020  RESULTS  S. Parameter Protocol Used Results Mg/I  Total Dissolved Solids(TDS)at 180°C  Total Hardness IS:3025 (Part 16)1984Reaffirmed)2017  Total Calcium (as Ca)  Calcium (as Mg)  Calcium (as Mg)  IS:3025 (Part 21)2009Reaffirmed)2014  April A 2320 B Part B Triation method Total Alkalinity April A 2320 B Part B Triation method Calcium (as Mg)  Solids(TDS)at 180°C  Protocol Used  RESULTS  Results Mg/I  Desirable Ilimit Max Permissible limit (in absence better alternate source)  2000  Im  Calcium (as Ca)  Calcium (as Ca)  IS:3025 (Part 21)2009Reaffirmed)2014  A 320  Calcium (as Mg)  IS:3025 (Part 49)991Reaffirmed)2014  A 320  Calcium (as Mg)  IS:3025 (Part 49)994Reaffirmed)2014  A 30.43  Do mm  APIA 2320 B Part B Triation method Total Alkalinity APIA 2320 B Part B Triation method Calcium (as Mg)  Solids (as Ci)  Solids (as Ci)  Solids (as Ci)  APIA 4500-F Part C ion selective method Less than 1  Total Calcium (as Ca)  B Nitrate (as NO3) APIA4500-No; Part D ion Mater electrode A 5.5 to 6.5 No relexation  B Nitrate (as NO3) APIA4500-No; Part D ion Mater electrode  (Second revision)  Notes:  The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory The test report can't be used for any publicity or any legal surpose.  The test report can't be used for any publicity or any legal surpose.  The test report can't be used for any publicity or any legal surpose.  The test samples meant for chamical analysis will be disposed off after 15 days from the date pf issue of test report unless until sequences.	Sample Location Date of Analysis started Date of Analysis started Date of Analysis completed  Of 10 2020  Dated of analysis completed  Of 10 2020  RESULTS  S. Parameter Protocol Used Results Mg/I  Total Dissolved Solids(TDS)at 180°C  Total Hardness IS:3025 (Part 16)1984Reaffirmed)2017  Total Calcium (as Ca)  Calcium (as Mg)  Calcium (as Mg)  IS:3025 (Part 46)1994Reaffirmed)2014  April A 2320 B Part B Trization method  Total Alkalinity APRIL 2320 B Part B Trization method Calcium (as Ci) Sincole (as Ci) Si		000 tello 000	Contract of the second			,	Sitelan of One	3
Sample received on dated Sample Location University Campus, Kurukshetra.  Date of Analysis started Dated of analysis completed  Of 10 2020  RESULTS  S. Parameter Protocol Used RESULTS  S. Parameter Protocol Used RESULTS  Total Dissolved Solida(TDS)at 180°C  IS 3025 (Part 16)1984Reaffirmed)2017  Solida(TDS)at 180°C  Total Hardness IS 3025 (Part 21)2009Reaffirmed)2014  A Calcium (as Ca)  Solida(TDS)at 180°C  Total Alkalinity APHA 2320 B Part B Tirabon method Chloride (as Cl) Si 3025 (Part 32)1988Reaffirmed)2014  APHA 4500-F Part C ion selective method RESULTS  No.  No.  Solida(TDS)at 180°C  Total Alkalinity APHA 4500-F Part C ion selective method RESULTS  APHA 4500-F Part C ion selective method RESULTS  No.  No.  Solida(TDS)at 180°C  So	Sample Location Date of Analysis started Dated of Analysis completed  Of 10 2020  Dated of Analysis completed  Of 10 2020  RESULTS  S. Parameter Protocol Used RESULTS  Results Mg/I  Desirable Ilimit Absence before a source before a sourc	Sample Location Date of Analysis started Dated of Analysis completed  Of 10 2020  RESULTS  S. Parameter Protocol Used Results Mg/I  Desirable Ilimit Absence before alternate source)  1 Total Dissolved Solids(TDS)at 180°C  IS 3025 (Part 16)1984Reaffirmed)2017 S56 S50 2000  Mg/I  2 Total Hardness IS 3025 (Part 21)2009Reaffirmed)2014 320 200 600 Mg/I  3 Calcium (as Ca) IS 3025 (Part 46)1994Reaffirmed)2014 Results Results Results Results Results Magnetium (as Mg/I  Absence before a source before a		DISTRIBUTED		17.70.70.70.70.70.70.70.70.70.70.70.70.70			519 S.D.Q 87	
Date of Analysis started   O6.10.2020	Sample Location Date of Analysis started Dated of analysis completed  Of 10,2020  RESULTS  S. Parameter Protocol Used Results No. Desirable Imit Max. Permissible limit absence before alternate source)  1 Total Dissolved Solids(TDS)at 180°C  2 Total Hardness IS:3025 (Part 16)1984Reaffirmed)2017 556 500 2000 rm  3 Calcium (as Ca) IS:3025 (Part 40)1981Reaffirmed)2014 78.15 75 200 rm  4 Magnesium (as Mg) IS:3025 (Part 46)1994Reaffirmed)2014 30.43 30 100 rm  5 Total Alkainity APHA 2320 B Part B Tiration method 310 200 600 rm  6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 67.35 250 1000 rm  7 Fluoride (as F) APHA 4500-F Part Clion selective method Less than 1 1.0 1.5 rm  8 Nitrate (as NO3) APHAA500-No; Part D Ion Meter electrode 0.6 45 45 rm;  9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relexation PHED Lab. Kurukshetra. Reliability of sample lis sender.  1 The results given above are related to the sample as received and testad in PHED Lab. Kurukshetra. Reliability of sample lis sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory  The test report can't be used for any publicity or any legal purpose.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date pf issue of test report unless until requested by the customer for retaining over a longer period.	Sample Location Date of Analysis started Dated of analysis completed  Of 10,2020  RESULTS  S. Parameter Protocol Used Results No. Parameter Results No. Parameter Protocol Used Results No. Parameter Results of the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample lie sender.  The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample lie sender.  The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory The test report can't be used for any publicity or any legal purpose.  The test report can't be used for any publicity or any legal purpose.  The test samples meant for chemical analysis will be disposed off after 15 days from the date pf issue of test report unless until a requested by the customer for retaining over a longer period.		100120000000000000000000000000000000000	TO THE REAL PROPERTY OF THE PARTY OF THE PAR			sami	-1110100	-
Dated of analysis completed   07.10.2020     RESULTS     S.   Parameter   Protocol Used   Results   Mg/l     Desirable   Max. Permissible limit (in absence better alternate source)     Total Dissolved   Solids (TDS) at 180°C   Solids (TDS) at 1	Dated of analysis completed   O7.10.2020   RESULTS	Dated of analysis completed   O7.10.2020   RESULTS					ukshetra.		7/10/200	20
S.   Parameter   Protocol Used   Results   IS:10500:2012(Second Revision)	S. Parameter Protocol Used Results Mg/I Desirable Is:10500:2012(Second Revision) Used No. Parameter Protocol Used Results Mg/I Desirable Isinit Max Permissible limit in absence better alternate source)  1 Total Dissolved Solids(TDS)at 180°C Is:3025 (Part 16)1984Reaffirmed)2017 556 500 2000 mg  2 Total Hardness Is:3025 (Part 21)2009Reaffirmed)2014 320 200 600 mg  3 Calcium (as Ca) Is:3025 (Part 46)1994Reaffirmed)2014 78:15 75 200 mg  4 Magnesium (as Mg) Is:3025 (Part 46)1994Reaffirmed)2014 30:43 30 100 mg  5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 mg  6 Chloride (as Cl) Is:3025 (Part 32)1988Reaffirmed)2014 67:35 250 1000 mg  7 Fluoride (as Cl) Is:3025 (Part 32)1988Reaffirmed)2014 67:35 250 1000 mg  8 Nitrate (as NO3) APHA4500-No; Part C ion selective method Less than 1 1.0 1.5 mg  9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH I	S. Parameter Protocol Used Results Mg/I Desirable Is:10500:2012(Second Revision) Used No. Parameter Protocol Used Results Mg/I Desirable Ising Max. Permissible limit (in absence better alternate source)  1 Total Dissolved Solids(TDS)at 180°C Is:3025 (Part 16)1984Reaffirmed)2017 556 500 2000 mg  2 Total Hardness Is:3025 (Part 21)2009Reaffirmed)2014 320 200 600 mg  3 Calcium (as Ca) Is:3025 (Part 46)1994Reaffirmed)2014 78:15 75 200 mg  4 Magnesium (as Mg) Is:3025 (Part 46)1994Reaffirmed)2014 30:43 30 100 mg  5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 mg  6 Chloride (as Cl) Is:3025 (Part 32)1988Reaffirmed)2014 67:35 250 1000 mg  7 Fluoride (as Cl) Is:3025 (Part 32)1988Reaffirmed)2014 67:35 250 1000 mg  8 Nitrate (as NO3) APHA4500-No; Part C ion selective method Less than 1 1.0 1.5 mg  9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH Is:3025 (Part 11)1983Reaffirmed)2017 7:72 6.5 to 8.5 No relexation 9 pH								
No.   Mg/I   Desirable   Max. Permissible limit (in absence better alternate source)	No.    Total Dissolved   IS:3025 (Part 16)1984Reaffirmed)2017   556   500   2000   make a source)	No.    Total Dissolved   IS:3025 (Part 16)1984Reaffirmed)2017   556   500   2000   mg		Dated	or analysis completed					-
Total Dissolved   IS:3025 (Part 16)1984Reaffirmed)2017   556   500   2000   r	Total Dissolved Solids(TDS)at 180°C  Total Hardness Solids(TDS)at 180°C  Solids(TDS)at 180°C  Total Hardness Solids(TDS)at 180°C  Total Hardness Solids(TDS)at 180°C  Solids(TDS)at 180°C  Solids(TDS)at 180°C  Total Hardness Solids(TDS)at 180°C  Solids(T	Total Dissolved Solids(TDS)at 180°C  Total Hardness Solids(TDS)at		S.	Parameter	Protocol Used	Results	IS:10500:20	112(Second Revision)	Ui
1 Total Dissolved Solids(TDS)at 180°C Solids(T	1 Total Dissolvad Solids(TDS)at 180°C IS:3025 (Part 21)2009Realfirmed)2014 320 200 600 ms  3 Calcium (as Ca) IS:3025 (Part 40)1991Realfirmed)2014 78.15 75 200 ms  4 Magneelium (as Mg) IS:3025 (Part 46)1994Realfirmed)2014 30.43 30 100 ms  5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 ms  6 Chloride (as Cl) IS:3025 (Part 32)1988Realfirmed)2014 67.35 250 1000 ms  7 Fluonde (as F) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 mg  8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D ion Meter electrode 0.6 45 45 mg  9 pH IS:3025 (Part 11)1983Realfirmed)2017 7.72 8.5 to 8.5 No relexation ph  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012 (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample is sender.  2 The test report can't be used for any publicity or any legal purpose.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of less use of test report unless until requested by the customer for retaining over a longer period requested by the customer for retaining over a longer period	1 Total Dissolvad Solids(TDS)at 180°C IS:3025 (Part 21)2009Realfirmed)2014 320 200 600 ms 3 Calcium (as Ca) IS:3025 (Part 40)1991Realfirmed)2014 78.15 75 200 ms 4 Magnesium (as Mg) IS:3025 (Part 46)1994Realfirmed)2014 30.43 30 100 ms 5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 ms 6 Chloride (as Cl) IS:3025 (Part 32)1988Realfirmed)2014 67.35 250 1000 ms 7 Fluonde (as F) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 mg 8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D ion Meter electrode 0.6 45 45 mg 9 pH IS:3025 (Part 11)1983Realfirmed)2017 7.72 8.5 to 8.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012 (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample is sender. 2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory. 3 The test report can't be used for any publicity or any legal purpose. 4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of lessue of test report unless until sequested by the customer for retaining over a longer period requested by the customer for retaining over a longer period		No.			Mg/I		absence better alternate	
3 Calcium (as Ca) IS:3025 (Part 40)1981Reaffirmed)2014 78:15 75 200 r 4 Magnesium (as Mg) IS:3025 (Part 46)1994Reaffirmed)2014 30:43 30 100 r 5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 600 600 600 600 600 600 600 60	3 Calcium (as Ca) IS:3025 (Part 40)1991Reaffirmed)2014 78:15 75 200 mm 4 Magnesium (as Mg) IS:3025 (Part 46)1994Reaffirmed)2014 30:43 30 100 mm 5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 mm 6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 67:35 250 1000 mm 7 Fluoride (as F) APHA 4500- F Part C ion selective method Less than 1 1:0 1:5 mm 8 Nitrate (as NO3) APHA4500-No <sub>1</sub> Part D ion Meter electrode 0:6 45 45 mm 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7:72 6:5 to 6:5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample lissender. 2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory. 3 The test report can't be used for any publicity or any legal purpose. 4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of test report unless until a requested by the customer for retaining over a longer period.	3 Calcium (as Ca) IS:3025 (Part 40)1991Reaffirmed)2014 78:15 75 200 mg 4 Magnesium (as Mg) IS:3025 (Part 46)1994Reaffirmed)2014 30:43 30 100 mg 5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 mg 6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 67:35 250 1000 mg 7 Fluoride (as F) APHA 4500- F Part C ion selective method Less than 1 1:0 1:5 mg 8 Nitrate (as NO3) APHA4500-No <sub>1</sub> Part D ion Meter electrode 0:6 45 45 mg 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7:72 8:5 to 8:5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list senider. 2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory. 3 The test report can't be used for any publicity or any legal purpose. 4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of less report unless until sequested by the customer for retaining over a longer period.		1		1S:3025 (Part 16)1984Reaffirmed)2017	556	500	2000	mg
3 Calcium (as Ca) IS:3025 (Part 40)1991Reaffirmed)2014 78:15 75 200 r 4 Magnesium (as Mg) IS:3025 (Part 46)1994Reaffirmed)2014 30:43 30 100 r 5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 600 600 600 600 600 600 600 60	3 Calcium (as Ca) IS:3025 (Part 40)1991Reaffirmed)2014 78.15 75 200 mg 4 Magnesium (as Mg) IS:3025 (Part 46)1994Reaffirmed)2014 30.43 30 100 mg 5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 mg 6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 87.35 250 1000 mg 7 Fluonde (as F) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 mg 8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D ion Meter electrode 0.6 45 45 mg 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 8.6 to 8.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra, Reliability of sample lissender.  2 The test report can't be used for any publicity or any legal purpose.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of test report unless until a requested by the customer for retaining over a longer period.	3 Calcium (as Ca) IS:3025 (Part 40)1991Reaffirmed)2014 78:15 75 200 mg 4 Magnesium (as Mg) IS:3025 (Part 46)1994Reaffirmed)2014 30:43 30 100 mg 5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 mg 6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 67:35 250 1000 mg 7 Fluonde (as F) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 mg 8 Nitrate (as NO3) APHA4500-No; Part D ion Meter electrode 0.5 45 45 mg 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7:72 8:5 to 8:5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra, Reliability of sample lissender.  2 The test report can't be used for any publicity or any legal purpose.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of test report unless until a requested by the customer for retaining over a longer period.		2	Total Hardness	IS:3025 (Part 21)2009Reaffirmed)2014	320	200	600	mg
4 Magnesium (as Mg) IS:3025 (Part 46)1994Reaffirmed)2014 30.43 30 100 1 5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 600 600 600 600 600 600 600 60	4 Magnesium (as Mg) IS:3025 (Part 46)1994Reaffirmed)2014 30.43 30 100 mg 5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 mg 6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 67.35 250 1000 mg 7 Fluoride (as F) APHA 4500-F Part C ion selective method Less than 1 1.0 1.5 mg 8 Nitrate (as NO3) APHA4500-No; Part D ion Meter electrode 0.6 45 45 mg 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes: 1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list sender. 2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory. 3 The test report can't be used for any publicity or any legal purpose. 4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of test report unless until a requested by the customer for retaining over a longer period.	4 Magnesium (as Mg) IS:3025 (Part 46)1994Reaffirmed)2014 30.43 30 100 mg 5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 mg 6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 67.35 250 1000 mg 7 Fluoride (as F) APHA 4500-F Part C ion selective method Less than 1 1.0 1.5 mg 8 Nitrate (as NO3) APHA4500-No; Part D ion Meter electrode 0.6 45 45 mg 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes: 1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list sender. 2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory. 3 The test report can't be used for any publicity or any legal purpose. 4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of lessue of test report unless until a requested by the customer for retaining over a longer period.		-		IS:3025 (Part 40)1991Reaffirmed)2014	78.15	75	200	mg
5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600  6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 67.35 250 1000 r.  7 Fluoride (as F) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 n.  8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D ion Meter electrode 0.6 45 45 m.  9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 8.5 to 8.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012 (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHEO Lab, Kurukshetra, Reliability of sample sender.	5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 ms 6 Chloride (as Ct) IS:3025 (Part 32)1988Reaffirmed)2014 67.35 250 1000 ms 7 Fluoride (as F) APHA 4500-F Part C ion selective method Less than 1 1.0 1.5 mg 8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D ion Meter electrode 0.6 45 45 mg 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relexation — 9 pH Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  Notes: 1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample is sender. 2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory. 3 The test report can't be used for any publicity or any legal purpose. 4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of test report unless until a requested by the customer for retaining over a longer period.	5 Total Alkalinity APHA 2320 B Part B Titration method 310 200 600 ms 6 Chloride (as Ct) IS:3025 (Part 32)1988Reaffirmed)2014 67.35 250 1000 ms 7 Fluoride (as F) APHA 4500-F Part C ion selective method Less than 1 1.0 1.5 mg 8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D ion Meter electrode 0.6 45 45 mg 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes: 1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list sender. 2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory. 3 The test report can't be used for any publicity or any legal purpose. 4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of lessue of test report unless until a requested by the customer for retaining over a longer period.					30.43			mg
6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 67.35 250 1000 7 7 Fluoride (as F) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 n 8 Nitrate (as NO3) APHA4500-No; Part D Ion Meter electrode 0.6 45 45 m 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relexation — Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012 (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra, Reliability of sample sericler.	6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 67.35 250 1000 ms  7 Fluoride (as F) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 mg  8 Nitrate (as NO3) APHA4500-No, Part D ion Mater electrode 0.6 45 45 mg  9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012 (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample is sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of less report unless until a requested by the customer for retaining over a longer period.	6 Chloride (as Cl) IS:3025 (Part 32)1988Reaffirmed)2014 67.35 250 1000 ms  7 Fluoride (as F) APHA 4500- F Part C ion selective method Less than 1 1.0 1.5 mg  8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D ion Meter electrode 0.6 45 45 mg  9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012 (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample is sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of less report unless until a requested by the customer for retaining over a longer period.		1 23			100000	757777	1707	mg
8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D Ion Meter electrode 0.6 45 45 m 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 8.5 to 6.5 No relexation — Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample sender.	8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D ion Meter electrode 0.6 45 45 mg 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 8.5 to 8.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS:10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample is sender.  1 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory.  The test report can't be used for any publicity or any legal purpose.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of test report unless until a requested by the customer for retaining over a longer period.	8 Nitrate (as NO3) APHA4500-No <sub>3</sub> Part D ion Meter electrode 0.6 45 45 mg 9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 8.5 to 8.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS:10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list sender.  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory.  The test report can't be used for any publicity or any legal purpose.  3 The test samples meant for chemical analysis will be disposed off after 15 days from the date of test report unless until a requested by the customer for retaining over a longer period.		-	Chloride (as Cl)					1
9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relexation —  9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relexation —  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012 (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab. Kurukshetra. Reliability of sample sender.	9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relevation  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample is sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory.  The test report can't be used for any publicity or any legal purpose.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of test report unless until a requested by the customer for retaining over a longer period.	9 pH IS:3025 (Part 11)1983Reaffirmed)2017 7.72 6.5 to 6.5 No relevation  Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample is sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory.  The test report can't be used for any publicity or any legal purpose.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of test report unless until a requested by the customer for retaining over a longer period.		7	Fluoride (as F)	APHA 4500- F Part C ion selective method		100		10000
Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHEO Lab. Kurukshetra. Reliability of sample sender.	Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample is sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of lessue of test report unless until a requested by the customer for retaining over a longer period.	Opinion: - The Parameter Tested at Sr.No1 to 9 in the test report meets the requirement of IS 10500:2012  (Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample lie sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of lessue of test report unless until a requested by the customer for retaining over a longer period.		8		744400000	7.72	8.5 to 8.5	No relexation	11000
(Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample sender.	(Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab. Kurukshetra. Reliability of sample list sender.  2 The test report can't be used for any publicity or any legal purpose.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of issue of test report unless until sequested by the customer for retaining over a longer period.	(Second revision)  Notes:  1 The results given above are related to the sample as received and tested in PHED Lab, Kurukshetra. Reliability of sample list sender.  2 The test report can't be regenerated/re-produced in whole or in part without written permission of Laboratory.  3 The test report can't be used for any publicity or any legal purpose.  4 The test samples meant for chemical analysis will be disposed off after 15 days from the date of issue of test report unless until sequested by the customer for retaining over a longer period.		9	pH	15:3025 (Part 11)1983Reallime0,2011	port monte t	he requirem	ent of IS 10500:2012	
4 The test samples meant for chemical analysis will be disputed on the samples meant for retaining over a longer period.	requested by the control	requested by the court			it and the ships of	bove are related to the sample as received a	ert without while	on permission c	, Los Tierre	
Sample analyzed by (Sheenu Sharma)  Sample analyzed by (Sheenu Sharma)  PHED Water Testing Laboratory, Kurykshetra  WHED Water Testing Laboratory, Kurykshetra  WHED Water Testing Laboratory  Regular  PHED Water Testing Laboratory  Regular  Regul	enest of CRE Build		Colored Colored	Wish 1	sender. The test report can The test report can The test samples or requested by the co	to be used to chemical analysis will be disposed of ustomer for retaining over a longer period on Sharma)	after 15 days	from the date p	rissue of test report unless	
sample analyzed by (Streens Sharma)  PHED Water Testing Laboratory, Kurykspelra  PHED Water Testing Laboratory, Kurykspelra  William CRE Building of Kole  PHED Water Testing Laboratory, Kurykspelra  Revious CRE  R	oriest of CRE Builter matters to. Shorter	militar out we we	ol o	Wish 1	sender. The test report can The test report can The test samples or requested by the co	the used to chemical analysis will be disposed of seant for chemical analysis will be disposed of statement for retaining over a longer period.  South of the season of th	phen W	from the date p	rissue of test report unless	
phen Water Testing Laboratory, Kurykshetra	const of CRE Builton when his Short Purposer	Comp forfor	6N/	Wish 1	sender. The test report can The test report can The test samples or requested by the co	the used to chemical analysis will be disposed of seant for chemical analysis will be disposed of statement for retaining over a longer period.  South of the season of th	phen W	from the date p	rissue of test report unless	











#### CHAPTER- 3 WASTE MANAGEMENT

#### 3.1 About Waste:

Human activities create waste, and it is the way these wastes are handled, stored, collected and disposed of, which can pose risks to the environment and to public health. Waste management is important for an eco-friendly campus. In University, different types of wastes are generated, the collection and management of which is very challenging.

Solid waste can be divided into three categories: bio-degradable, non-biodegradable and hazardous waste. A bio-degradable waste includes food wastes, canteen waste, wastes from toilets etc. Non-biodegradable wastes include what is usually thrown away in homes and schools such as plastic, tins and glass bottles etc. Hazardous waste is waste that is likely to be a threat to health or the environment like cleaning chemicals, acids and petrol.

Unscientific management of these wastes such as dumping in pits or burning them may cause harmful discharge of contaminants into soil and water supplies, and produce greenhouse gases contributing to global climate change respectively. Special attention should be given to the handling and management of hazardous waste generated in the University. Biodegradable waste can be effectively utilized for energy generation purposes through anaerobic digestion or can be converted to fertilizer by composting technology. Non-biodegradable waste can be utilized through recycling and reuse. Thus the minimization of solid waste is essential to a sustainable University. The auditor diagnosed the prevailing waste disposal policies and suggested the best way to combat the problems.

Table 3.1 Different types of waste generated in the University Campus.

Sr. No.	Types of Waste	Particulars
1	Solid wastes	Damaged furniture, paper waste, paper plates, food wastes etc.
2	Plastic waste	Pen, Refill, Plastic water bottles and other plastic containers, wrappers etc.
3	E-Waste	Computers, electrical and electronic parts etc.
4	Glass waste	Broken glass wares from the labs etc.
5	Chemical wastes	Laboratory waste etc.
6	Bio-medical Waste	Sanitary Napkin etc.





#### 3.2 Waste management Practices adopted by the University

♣ University has adopted "One dust Bin" waste collection system. All kinds of waste generated from various activities is collected in single bin and all the dustbins are filled in a trolley and collected at one place.





Figure 3.1: - Waste collection bin in KUK University





#### **Recommendation:**

It is recommended to adopt 3 Bin Waste Collection System for collection of different types of waste generated in University premises.



Figure 3.2: - Recommended 3 Dust Bin waste collection System

#### 3.3 Waste Collection Points in University: -

Audit team also visited various departments, canteen, and residential area to find out waste generation area and waste collection points for further improvement. Details are given in the table.

Table: 3.2 Details of Waste collection Dust bin system in KUK University

Sr. No	Area	Dust Bin (Type )	No of Dust Bin
1	Girls Hostel	Big (Stationary Type)	10
2	Girls Hostel	Small type (Portable)	09
3	Boy Hostel	Big (Stationary Type)	10
4	Boy Hostel	Small type (Portable)	10
	University market and		
5	Residential area	Big (Stationary Type)	12
	Near Departments in KUK		
6	campus	Big (Stationary Type)	14
7	Gardens Area	Small type (Portable)	130
	Total	Area	195





#### 3.4 Organic Waste Generation in University: -

♣ Audit Team also visited in sanitization department and discussed with sanitary officer about the waste collection process. University has approx. 2.5 Ton per day waste generation. Department wise Generated organic waste are as follows: -

Table: 3.3 Detailed of Organic Waste Generation in KUK

Sr. No	Building Name	Approx. Quantity (Kg)
1	Bheema Bhawan	25
2	Dr. B.R. Ambedkar Bhawan	30
3	Subhash Bhawan	27
4	Pratap Bhawan	35
5	Ch. Ranbir Singh Hostel	27
6	Narhari Bhawan	22
7	Ch. Devilal Bhawan	27
8	Shaheed Bhagat Singh Bhawan	28
9	Tagore Bhawan	30
10	Arjun Bhawan	26
11	Vivekanand Bhawan	30
12	Harsh Bhawan	28
13	Girls Hostel-1 (Saraswati Bhawan)	30
14	Girls Hostel-2 (Meera Bhawan)	27
15	Girls Hostel-3 (Kasturba Bhawan)	37
16	Girls Hostel-4 (Bharti Bhawan)	27
17	Girls Hostel-5 (Gargi Bhawan)	34
18	Girls Hostel-6 (Subhadra Bhawan)	35
19	Girls Hostel-7 (Ahilya Bhawan)	26
20	Girls Hostel-8 (Ganga Bhawan)	28
21	Girls Hostel-9 (Uttra Bhawan)	36
22	Girls Hostel-10 (Devyani Bhawan)	38
23	Girls Hostel-11 (Kalpna Chawla)	27
24	Girls Hostel-12 (Laxmi Bai)	30
25	Girls Hostel-13 (Yamuna Bhawan)	28
26	International Guest House	10
	Total	748

**Note:** -All Data are collected on Dated 10.03.2021 and 11.03.2021. At present all hostel are working 40 % to 60 % Strength due to covid- 19. The Waste quantity generation depends on occupancy of the Students.





#### Recommended: Install Organic waste composting Machine in University: -

An organic waste composting machine is an independent unit that facilitates the composting process and provides better composts. It takes waste as its input and provides manure as its output. Composting without an organic waste composting machine will take a considerable amount of time.









#### **About Composting Process: -**

Highly compact composting machine uses special microorganisms to break down and decompose all kinds of organic waste into compost within 24 hrs with a volume reduction of 85-90%. When organic waste is added to it, moisture is sensed by the humidity sensor, heater, mixing blades and an exhaust system.



#### **Recommendation: -**

University has a good potential to install organic converter.





# CHAPTER- 4 RAIN WATER HARVESTING SYSTEM

#### 4.1. Rain water harvesting systems

The rainwater harvesting is a technique to capture the rainwater when it precipitates, store that water for direct use or charge the groundwater and use it later.

There are typically four components in a rainwater harvesting system:

- Roof Catchment.
- **♣** Collection.
- **4** Transport.
- **♣** Infiltration or storage tank and use.

If rainwater is not harvested and channelized, it runoffs quickly and flows out through stormwater drains. For storm-water management, the recharge pits, percolation pits and porous trenches are constructed to allow storm water to infiltrate inside the soil.

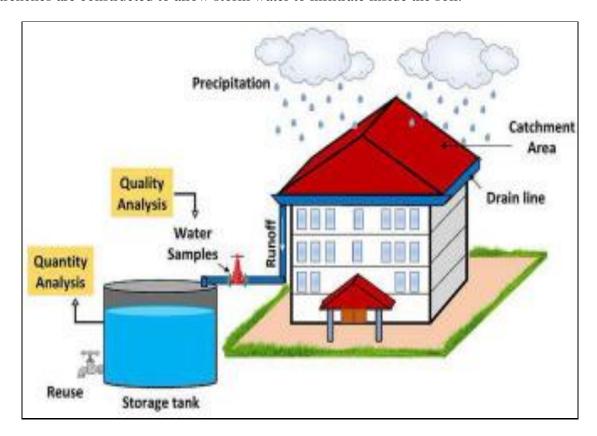


Figure: -4.1 Components of a rooftop rainwater harvesting system



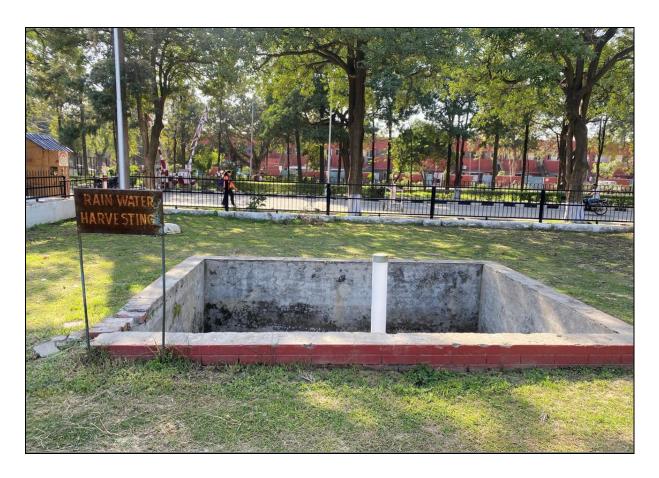


#### 4.2 Rainwater harvesting system in KUK University:

University has installed 14 no of rain water harvesting systems and it is working by method of roof top rain water harvesting and by constructing recharge pit.

Table 4.1: -Details of RWH system in KUK:-

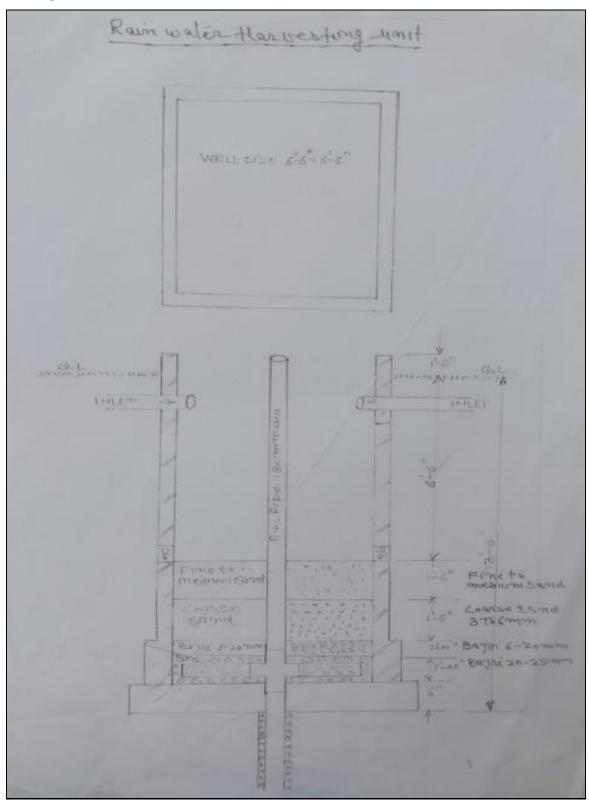
Sr. No	Area Name	Pit Quantity
1	University 2nd Gate (Near Toilet Block in Old Law)	1
2	3 rd Gate (MTH House Side)	1
3	RK Sadan Lawn	1
4	Geology Department	1
5	Uttra Bhavan (Girls Hostel No-09)	2
6	Near Ganga Bhavan (Girls Hostel - 08)	1
7	Challis Bag	1
8	Near Arjun Bhavan	1
9	Dharohar Phase- II	1
10	Exam-III	2
11	Shooting Range	2
	Total	14







#### 4.3 Design of RWH in KUK:-







#### 4.4 Photograph of RWH in KUK:-





#### **4** Observation :-

University has 14 Nos RWH systems that are installed at varius locations and additional units are also proposed. **Its Appricable**. There is good potential to install RWH system in reaming area in University.





# END OF THE REPORT THANKS